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Tinoco, Sauer, Wang, Puglisi, Harbison & Rovnyak, Physical ...
The reaction is obviously acid catalyzed (since H+ affects the order but does not appear as a reactant) This mechanism in the steady state approximation gives the correct rate law: ketone + H+→H+· ketone (slow) H+· ketone+ I 2 → i odoketone + 2H++ I- (fast) a) The overall order is 2.

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